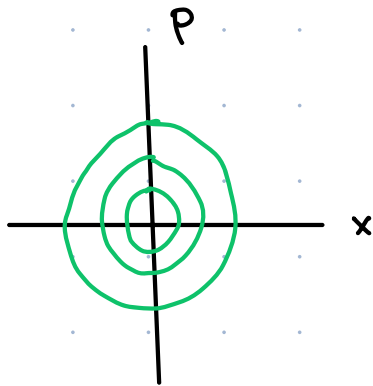
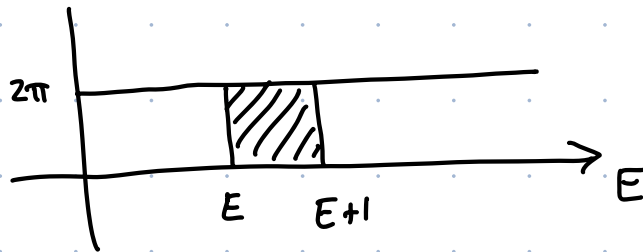


Last time: density of states by energy for Harmonic oscill.



$$H = \frac{1}{2} (x^2 + p^2)$$

$$\begin{aligned} H_*(dp \wedge dx) &= H_*(dH \wedge d\theta) \\ &= 2\pi dH \end{aligned}$$



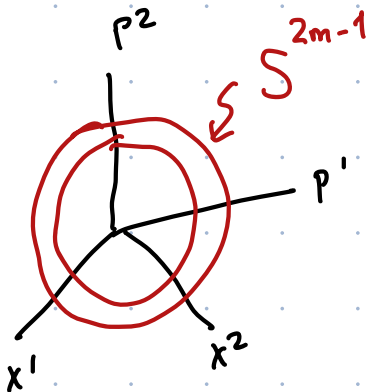
Combining multiple copies of Harmonic osc :

$$X(m) = \mathbb{R}^m$$

m copies of HO

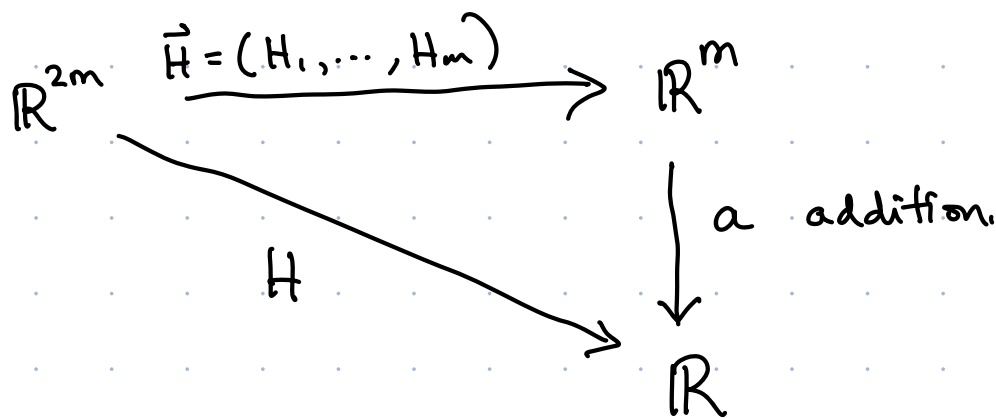
$$M(m) = (T^*\mathbb{R})^m = \mathbb{R}^m \times \mathbb{R}^{m*}$$

$$H = \sum_{i=1}^m H_i = \frac{1}{2} \left(\sum_{i=1}^m x_i^2 + \sum_{i=1}^m p_i^2 \right)$$



energy level sets are spheres of \dim^{2m-1}
 \Rightarrow much faster growth of volume
 of state space as a f^n of energy

Compute $H_*(\mathcal{V}_{\text{Liouville}}) = \frac{\omega^m}{m!} = dp_1 \wedge \dots \wedge dp_m \wedge dx^1 \wedge \dots \wedge dx^m$



$$\vec{H}_*(\mathcal{V}_{\text{Liouville}}) = \begin{cases} (2\pi)^m \cdot 1 & \text{in positive corner} \\ 0 & \text{else} \end{cases} \quad E_\varepsilon \geq 0$$

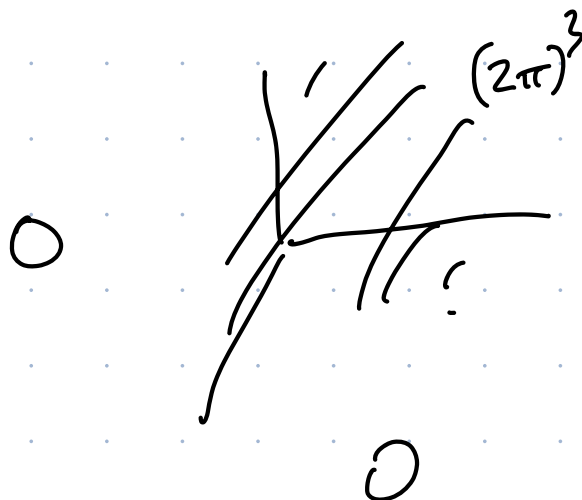
$m=1$



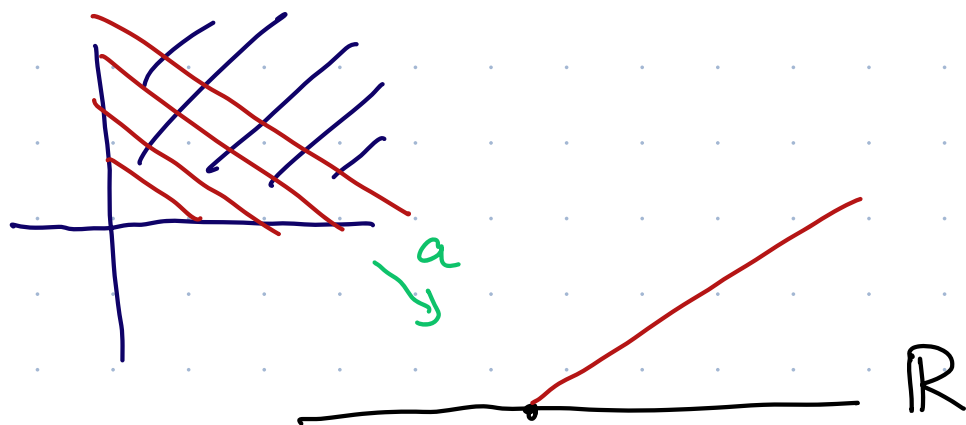
$m=2$



$m=3$



applying sum: $a_*(\vec{H}_* \text{vol}_{\text{Liouville}})$



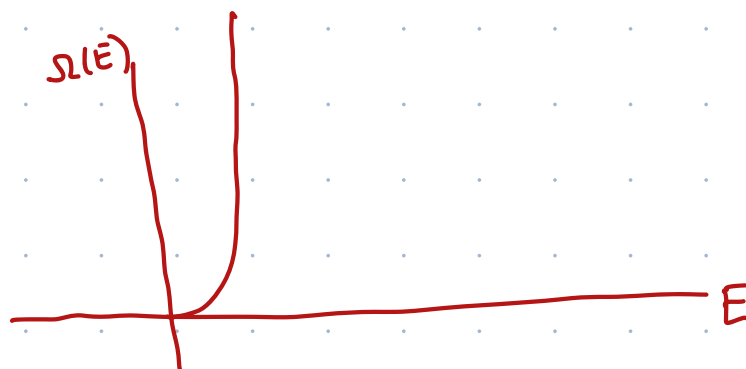
$m=2$

$$\Omega(E)dE = C_2 E dE$$

linear

$$m=3 \quad \Omega(E)dE = C_3 E^2 dE$$

in general for m oscillators $\Omega(E)dE = C_m E^{m-1} dE$



Def: Entropy of system, as a f^n of total energy:

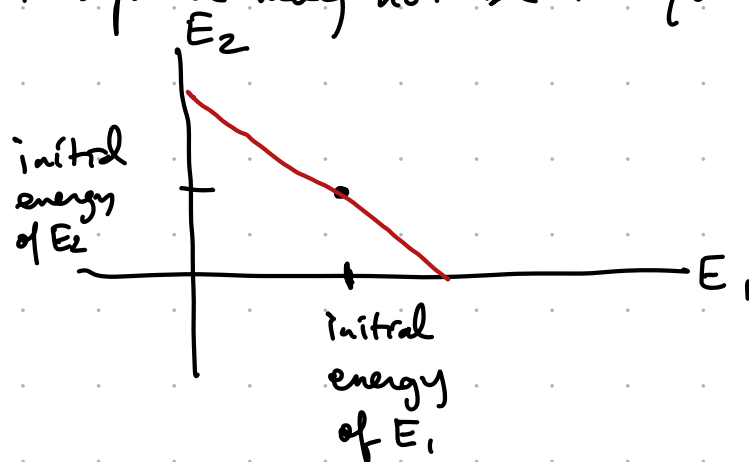
$$S(E) = \log \Omega(E)$$

Temperature & Boltzmann distribution.

Idea:

i) Equipartition of energy
when system is at equilibrium,
states of equal energy are equally likely.

ii) If two systems (M_1, ω_1, H_1) , (M_2, ω_2, H_2)
are each in equilibrium,
The joint system may not be in equilibrium



$$E_{\text{tot}} = E_1 + E_2$$

Warning: Without interaction terms in $H_{\text{tot}} = H_1 + H_2$,
 H_1 and H_2 are separately conserved:

$$\{H_{\text{tot}}, H_1\} = \{H_{\text{tot}}, H_2\} = 0$$

Implicit in Equipartition for joint systems
is that there are small unknown interaction
terms \Rightarrow we may wait longer for equilibrium.
(making the Hamiltonian system "Ergodic").

Main question: How to determine equilibrium of Joint system
i.e. at equilibrium, what are energies of systems (4.11)?

Aside: $(\mathbb{R}^{2n}, \nu_{\leftarrow})$ volume form
 $\downarrow H$
 \mathbb{R}

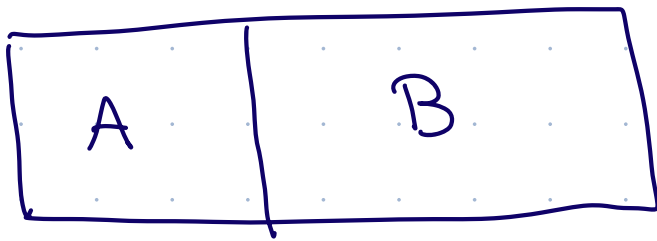
\Rightarrow level sets inherit volume forms

$$H^{-1}(c) = \sum_c \int \nu_c \in \Omega^{\text{top}}(\Sigma_c)$$

$$\text{st. } \nu_c \wedge dH = \nu_{\text{Liou}}|_{H^{-1}(c)}$$

ν_c defines uniform dist. on states of fixed energy

(assuming $H^{-1}(c)$ compact)



two systems, A + B
joined together

density of states: $\Omega_A(x)$, $\Omega_B(x)$
by energy

total energy constant: $x + y = E_{\text{tot}}$

Q.: what is prob. that system A has energy x
(and that syst. B has energy $E_{\text{tot}} - x = y$)

$$P(H_A = x) dx$$

$P \propto$ number of states of joint system
w/ Energy E_{tot} st. $H_A = x$

$$P \propto \Omega_A(x) \Omega_B(E_{\text{tot}} - x)$$

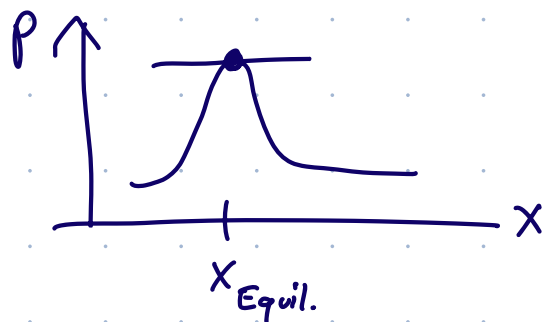
Q.: for what value of x is P maximum?

i.e.

$$\left. \frac{\partial P}{\partial x} \right|_x = 0$$



$$\left. \frac{\partial \log P}{\partial x} \right|_x = 0$$



i.e.
$$\left. \frac{\partial \log \Omega_A(x)}{\partial x} \right|_{x_{eq}} + \left. \frac{\partial \log \Omega_B(E_{tot} - x)}{\partial x} \right|_{x_{eq}} = 0$$

i.e.
$$\left. \frac{\partial \log \Omega_A}{\partial x} \right|_{x_{eq}} = \left. \frac{\partial \log \Omega_B}{\partial y} \right|_{y = E_{tot} - x}$$

equality of derivatives of entropies
wrt to subsystem energy.

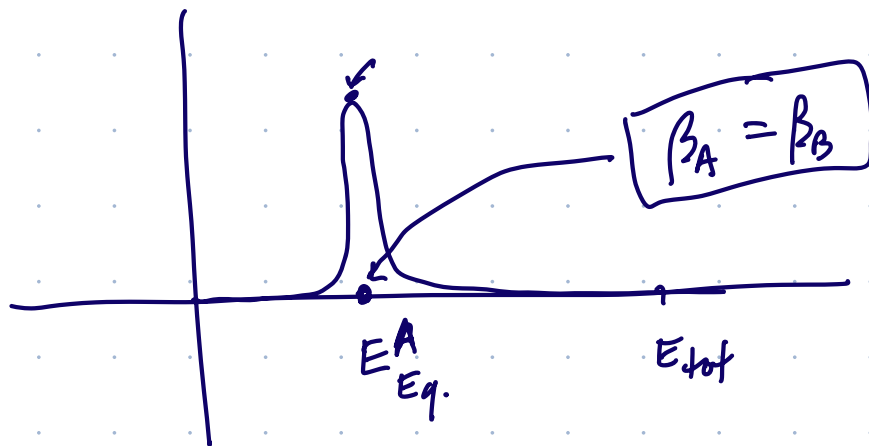
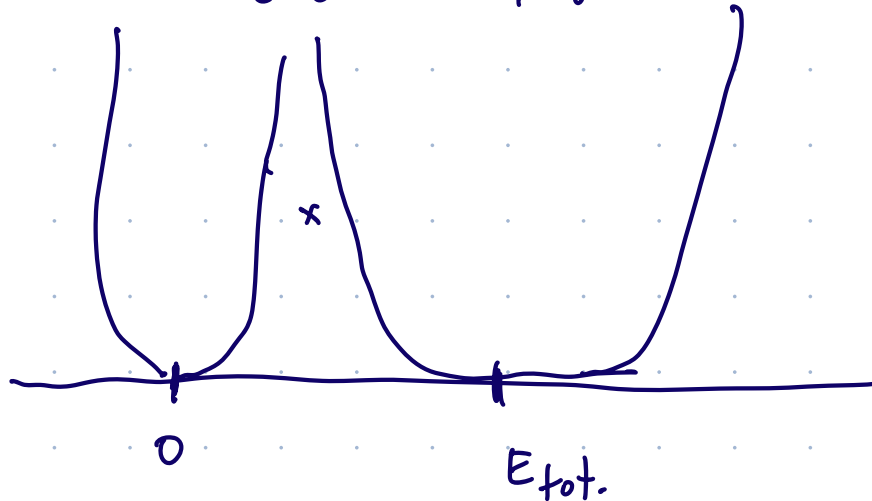
Def: The inverse temperature $\beta \left(= \frac{1}{kT} \right)$
is this quantity $T =$
temp.

$$\beta = \frac{\partial \log \Omega(E)}{\partial E}$$

"Two subsystems are in equilibrium when they
have same temperature"

e.g. $\left. \begin{aligned} \Omega_A(E) &= C_A E^m \\ \Omega_B(E) &= C_B E^n \end{aligned} \right\}$

$$P \propto E_A^m (E_{\text{tot}} - E_A)^n$$



Remark: β can be negative for systems where energy bounded above, in this case as β passes through zero

$T \rightarrow +\infty = -\infty \rightarrow$ negative values

Boltzmann distribution

Pair of systems A, B $E_A \ll E_B$

("B is a heat bath" or "reservoir")

Assume (A, B) is in equilibrium.

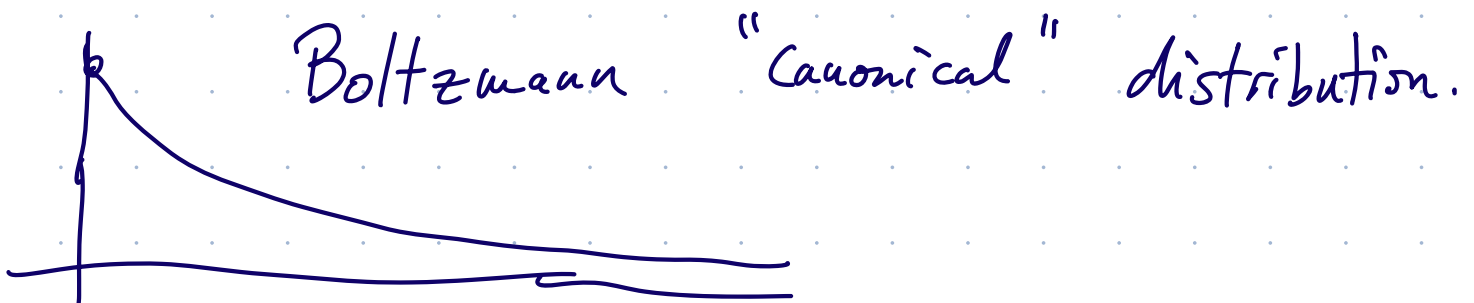
Q.: What is prob. that A is in a specific state of energy E_A

$$P \propto \Omega_B(E_{\text{tot}} - E_A)$$

$$\log \Omega_B(E_{\text{tot}} - \underbrace{E_A}_{\text{small}}) \stackrel{\text{Taylor}}{=} \log \Omega_B(E_{\text{tot}}) - \left(\frac{\partial \log \Omega_B}{\partial E_B} \right)_{E_{\text{tot}}} E_A + \dots$$

$$\Rightarrow \boxed{\Omega_B(E_{\text{tot}} - E_A) = \Omega_B(E_{\text{tot}}) e^{-\beta_B E_A} + \dots}$$

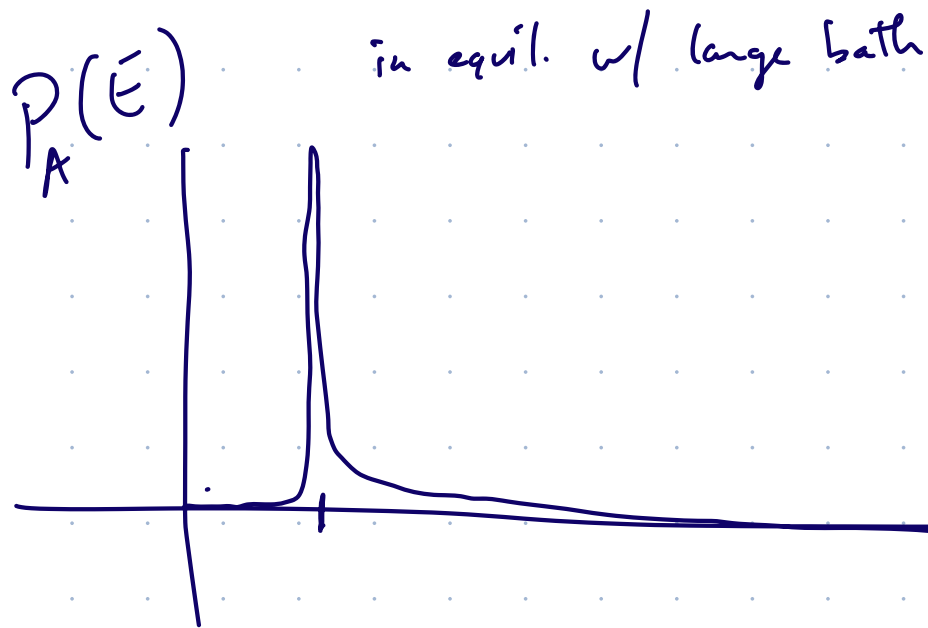
$$\Rightarrow \boxed{P(A \text{ has specific state of energy } E_A) = C \cdot e^{-\beta E_A}}$$



⇒ obtain Prob. density for E_A :

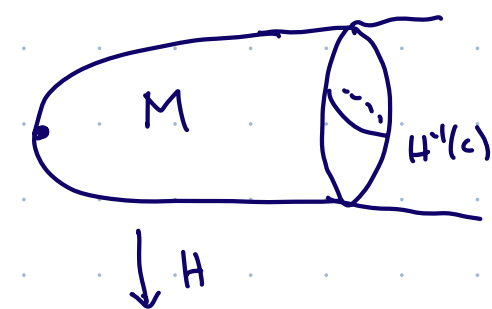
$$P_A(E) dE = C \cdot \Omega_A(E) e^{-\beta E} dE$$

Book: (Souriau : structure of
dynamical systems.
English by Cushman et al.)



Push-forward:

$$H: M \rightarrow \mathbb{R} \quad C^\infty$$



- Proper ($H^{-1}([a,b])$ compact).
(e.g. $\sum (x_i^2 + p_i^2)$)

- $c \in \mathbb{R}$ regular value

(dH non-vanishing on $H^{-1}(c)$).

- v volume form on M

$$H_* v = \int_{H^{-1}(c)} v$$

to define this: in a nbhd of $H^{-1}(c)$,

$$v = \underset{\substack{\uparrow \\ \text{normal}}}{dH} \wedge \underset{\substack{\uparrow \\ \text{Longitudinal}}}{u} \quad u \in \Omega^{2n-1}(\mathcal{U})$$

for each $H \in (c-\varepsilon, c+\varepsilon)$ define

$$\boxed{\Omega(H) = \int_{H^{-1}(c)} u}$$

← requires $H^{-1}(c)$ to be oriented

then $H_* v = \Omega(H) dH$

(we use natural

More generally:

orient. defined
by u)

$$\Omega(H) \geq 0$$

$$\begin{array}{c} E \\ \downarrow \pi \\ M \end{array}$$

fibre bundle

w/ compact oriented fibres
of dim k

$$\pi_* : \Omega^p(E) \rightarrow \Omega^{p-k}(M)$$

"fiber integration"